

### Total Synthesis of a $\gamma$ -Carboxymethyltetronic Acid. (*S*)-Carlosic Acid

**Summary:** The first total synthesis of a naturally occurring mold tetronic acid with correct absolute configuration is described, as well as a possible biogenetic precursor for the entire family.

**Sir:** In recent biosynthetic studies,<sup>1</sup> we demonstrated that carlosic acid (1) was the major precursor of (*R*)-carolic acid (2) in *Penicillium charlesii*. We noted that no synthetic work had been done on any of the mold tetronic acids bearing the  $\gamma$ -carboxymethyl substituent. Furthermore, the reported total syntheses in the  $\gamma$ -methyl series, viz., ( $\pm$ )-carolic acid<sup>2</sup> and ( $\pm$ )-carolinic acid,<sup>3</sup> were not applicable to either work with chiral compounds or isotopic labeling.<sup>4</sup> This communication describes the first example of a total synthesis of a mold tetronic acid in its correct absolute configuration and incorporates all of the desirable features described above.

The key step in the synthesis involved the cyclization of 3, which was formed in 80% yield from dimethyl (*S*)-malate and diketene (Et<sub>3</sub>N catalyst, PhH). The nmr spectrum of 3 was similar to the starting ester. In addition to the malate moiety [ $\delta$  3.67 (3 H, s, ester), 3.72 (3 H, s, ester), 2.90 (2 H, d,  $J$  = 6 Hz, methylene), and 5.47 (1 H, t,  $J$  = 6 Hz, methine)], new signals appeared at  $\delta$  2.25 (3 H, s, acetyl) and 3.50 (2 H, s, methylene) (CDCl<sub>3</sub>) for the acetoacetyl group. Compound 3 was very thermolabile, and had to be purified by chromatographic means (alumina). The cyclization of 3 to 4 had to be carried out at a low temperature; otherwise mainly dimethyl fumarate was obtained (with concomitant loss of CO<sub>2</sub> and acetone). Treatment of 3 with *t*-BuOK in *t*-BuOH at the freezing point effected a 39% yield of 4 in which the acetoacetyl methylene signal and the ester signal at  $\delta$  3.72 were no longer present. In addition to nmr signals at  $\delta$  2.38 (3 H, s, acetyl), 3.67 (3 H, s, ester), 2.72 (2 H, m, methylene), and 4.57–4.75 (1 H, m, methine), a new signal appeared at  $\delta$  8.42 (1 H, s, enol) (CDCl<sub>3</sub>). The bromination of 4 to 5 had to be carried out rapidly owing to the sensitivity of the ester function to HBr liberated by the reaction. Com-

pound 5 had a similar nmr spectrum to 4 except for loss of the acetyl signal ( $\delta$  2.38) (DMSO-*d*<sub>6</sub>). Its structure was confirmed by conversion to the free carboxylic acid which had been obtained from carlosic acid by degradation.<sup>5</sup> The catalytic reduction of 5 to 6 [which had a nmr similar to 5 except for the appearance of a new signal at  $\delta$  4.99 (1 H, s, vinylic) (CDCl<sub>3</sub> + 5% DMSO-*d*<sub>6</sub>)] was carried out similarly to that for  $\alpha$ -bromo-(*S*)- $\gamma$ -methyltetronic acid.<sup>6</sup>

Excepting the cyclization, all synthetic yields were in the 70–80% range. Elementary analyses and spectral data for all of the above compounds were in agreement with the assigned structures.

Since our biosynthetic studies<sup>1</sup> seemed to indicate that *P. charlesii* contained a relatively nonspecific biological acylation system, the compound 6 represents a potential intermediate in both the biosynthesis (as the free acid) and synthesis of carlosic acid (1), carlic acid (7), and viridic acid (8). In the specific instance of carlosic acid (1), treatment of 6 with butyryl chloride, TiCl<sub>4</sub>, and PhNO<sub>2</sub> gave the ester 9, which was converted by gentle saponification to 1. The 1 thus obtained was identical in all respects with the natural product. The application of intermediate 6 to the synthesis of 7 and especially 8 should be straightforward. Our present work allows specific isotopic labeling of 9 or 1 via use of PrC\*OCl, which is available with either <sup>14</sup>C or <sup>13</sup>C label as shown.

A full account will be given of this work upon completion.

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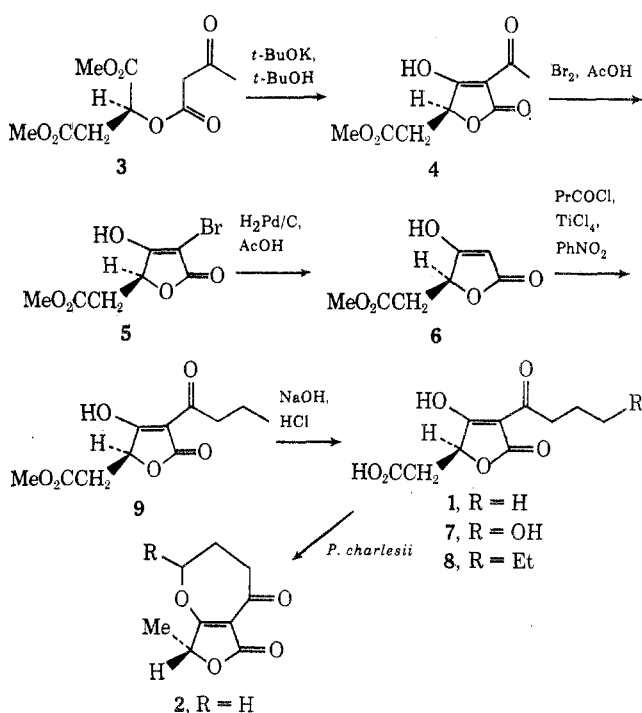
### References and Notes

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### Nucleic Acid Related Compounds. 9. The Synthesis of 6-Amino-9-(2-deoxy-D-erythro-pent-1-enofuranosyl)- purine, the First 1',2'-Unsaturated Purine Nucleoside<sup>1,2</sup>

**Summary:** Adenosine has been transformed into 6-amino-9-(2-deoxy-D-erythro-pent-1-enofuranosyl)purine (4b) by elimination of hydrogen iodide from a suitably blocked 2'-iodo derivative, and hydrogenation of 4b completes the conversion to  $\alpha$ - and  $\beta$ -2'-deoxyadenosines.

**Sir:** Access into unsaturated pentofuranosyl nucleosides including the 2',3',<sup>3</sup> 3',4',<sup>4</sup> and 4',5'<sup>5</sup> olefinic systems has been reported. However, no authenticated 1',2'-unsaturated purine nucleoside has been described, although the antibiotic augustmycin A (decoyinine)<sup>6</sup> was originally assigned this structural feature.<sup>7</sup> It has been considered that biological transformation of ribo nucleosides to their 2'-